# focus on Mass Spectrometry & Spectroscopy

# **Direct Determination of Trace Elements** in Body Fluids Using ICP-Mass Spectrometry

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The better understanding of the relevance of various elements in metabolic and other processes in organisms leads to increased interest in the analysis of biological samples like body fluids. Here, the focus is on supply of essential elements and detection of toxic elements. Fast and more sensitive analytical techniques are necessary to satisfy these demands.

Inductively Coupled Plasma Mass Spectrometry offers low detection limits, multi-element capabilities, simplified sample preparation and low sample consumption. It is very useful for the determination of trace and ultra-trace elements in various biological samples.

## Analysis of Various Matrices

The analysis of biological samples allows a variety of conclusions about the current status of an organism

Different matrices can be investigated (Table 1). The individual properties of the body fluids are related to specific functions in the organism. Therefore the selection of the right matrix is crucial to get the correct answer for a specific question.

Table 1. Sample matrix body fluids

Research area	Appropriate sample matrix	
Medicine		
Monitoring of the mineral balance	Urine	
Detection of medicines and addictive drugs	Plasma	
• Exposition to harmful substances	Serum	
Detection of toxins	Whole blood	
Historical samples	Hair	
Therapeutic medicine		
Pharmacodynamics		

Blood is a transport medium. The trace element concentration depends on short term uptake. Since an insufficient mineral uptake is balanced with body's own reserves e.g. serum cannot indicate mineral deficiencies.

Heavily discussed is the value of hair analyses. Results are highly dependent on age, sex, length and colour as well as care products and environmental factors. The endogenous and exogenous portions cannot be distinguished and a correlation of elemental concentrations in hair and blood/urine is not proven [1]. Generally accepted is the analysis of hair samples for the detection of addictive drugs and the analysis of ancient samples. [2]

# Analytical Methods

Traditionally the characterisation of body fluids was performed using atomic absorption techniques. The combination of flame (F-AAS) and graphite furnace AAS (GF-ASS) is able to cover a large concentration range. Since AAS is a single element technique and needs an individual excitation source the analysis is time consuming and the number of elements, that can be characterised, is limited

Interferences that are formed by the sample matrix and e.g. Argon or Oxygen can be removed with interference management systems. These systems use Hydrogen and Helium to generate collisions and reactions with the molecular interferences. As a result, new and non-interfering species are formed or the kinetic energy of the interfering molecules is decreased so that they do not reach the mass filter.

The sample preparation is very simple. All liquid matrices can be diluted and directly analysed with the ICP-MS. The calibration of the method can be performed using an external calibration with different calibration levels or by using a standard addition calibration on a real sample. The standard addition calibration is useful if matrix effects will influence the sample introduction and excitation in the plasma.

## Instrumentation

Direct multi-element analysis of plasma and whole blood control materials was carried out using an ICP-MS, the PlasmaQuant® MS. All work was done under routine analytical laboratory conditions, not 'clean room' conditions.



## Materials and Reagents

High purity nitric acid (Baseline<sup>®</sup>, Seastar Chemicals), Triton-X 100 (Sigma Aldrich) and deionised water (18 M $\Omega$  cm<sup>-1</sup>) were used in the preparation of sample and calibration solutions. All labware, new or used, was thoroughly cleaned by acid washing and rinsing, and then the clean containers were left filled with 2% v/v HNO<sub>3</sub> until use. Three multielement calibration solutions are prepared from a multielement solution in 2% v/v HNO<sub>3</sub>. An internal standard solution, containing 10  $\mu$ g/L of Sc, Rh and Ir, was prepared with 1% v/v HNO... The internal standard was added to the nebuliser through a 'Y-piece'.

Figure1. PlasmaQuant® MS Elite

With ICP-MS a multi-element technique is nowadays used since it offers the low detection limits of GF-AAS and a large dynamic range of 10 orders of magnitude. A characterisation that took between 30 sec and 5 min for just one element is now possible for >20 elements in less than 5min

ICP-MS, with its capability to detect individual isotopes, can be furthermore applied for long term studies with isotopic enriched medical drugs.

## Method Development

For the analysis of whole blood the challenges are the sample matrix and interferences that influence the analysis for elements of interest. The primary focus is on essential but also on toxic elements like Selenium, Arsenic, Cadmium, Lead and Chromium.

## Sample Analysis

All certified materials were prepared according to the manufactures instructions. After carefully dissolving the materials the samples were diluted with a diluent of 0.5% v/v HNO3 und 0.005% v/v TritonX-100.

The measured values shown are the average of two repeat measurements.

The reference material ClinChek® plasma control Level 1 and 2 (Recipe®) was dissolved in 3 mL deionised water and subsequently diluted 1:10 using the diluent solution. The obtained results (Table 2) are in perfect match with the certified concentrations.

The reference material 'trace elements in whole blood' (Seronorm™) was diluted 1:20 after carefully dissolving it in 5mL deionised water. As seen from Table 3, excellent agreement between the measured and the certified values is observed.

#### **INTERNATIONAL LABMATE - APRIL 2015**

Element		Plasma Level 1		Plasma Level	Plasma Level 2	
		Measured	Certified	Measured	Certified	
Cd	µg/l	2.2	2.0 - 3.4	9.8	9 - 15	
Cr	µg/l	3.5	2.6 - 4.2	14	11 - 16	
Со	µg/l	5	4,1 - 6.7	16.6	14 - 22	
Cu	µg/l	850	634 – 1056	1290	1050 - 1750	
Fe	µg/l	756	563 - 937	964	859 – 1431	
Li	mg/l	2.4	1.9 – 3.1	5.2	3.9 – 6.5	
Mg	mg/l	26	23 - 37	33	32 – 37	
Mn	µg/l	4.9	3.8 – 6.2	14.6	11 - 17	
Мо	µg/l	1.1	0.7 – 1.2	6.6	4.5 – 7.5	
Ni	µg/l	7.4	5.9 – 9.7	18	16 - 26	
TI	µg/l	0.03	<1	2.5	2.1 – 3.5	
Zn	µg/l	1113	823 - 1371	1338	1133 - 1887	

Table 2. Results for the analysis of ClinChek®-plasma control Level 1 and 2 (Ch.-B.: 417)

### Conclusion

This work has successfully demonstrated that the ICP-MS PlasmaQuant® MS provides a simple and very effective solution for the direct determination of trace elements in complex samples such as plasma and whole blood. The fast multielement capability and low detection limits will increase the use of ICP-MS in the characterisation of body fluids. Easy handling and straightforward software solutions have increased the use of this method in the past years.

# 4 Great Things About Flame Spectrometers

Imagine a miniature spectrometer that gives you the familiar benefits of modular spectroscopy – without compromise. Flame is designed with the performance-enhancing features you've asked for and produced using industry-leading manufacturing techniques refined from decades of experience. Here are four great reasons to consider Flame for your next application:

### 1. Flame spectrometers are highly configurable.

With Flame, users can select among detectors, optical bench accessories and filter options to configure systems for thousands of applications. Its interchangeable slit design lets users adjust resolution and throughput on demand. Switch between absorbance and fluorescence measurements with ease.

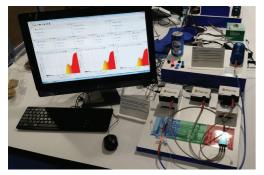
#### 2. Flame has great thermal stability.

Advances in optical design and assembly processes have resulted in a spectrometer with thermal stability of 0.05 nm/°C over a 650nm range. This is significant for process and similar applications where maintaining measurement consistency in a changing environment is critical.

#### 3. Flame is produced with low

unit to unit variation.

New manufacturing techniques that combine the best of automated processes with the artistry of precision assembly have dramatically improved spectrometer to spectrometer reproducibility. This benefits OEMs and integrators who appreciate the small size and great value of miniature spectrometers, but require highly consistent performance from unit to unit.



Multipoint absorbance sampling using three Flame



Element		Seronorm Level 1		Seronorm Level 2	
		LOT 404107		LOT MR9067	
		Measured	Certified	Measured	Certified
Cd	µg/l	0.72	0.67 -0.76	5.77	5.4 – 7.2
Со	µg/l	0.13	<1	5.3	5.2
Cr	µg/l	1.3	1.2	7.2	7.1
Mn	µg/l	9.5	9	13.9	12.8 – 15.1
Ni	µg/l	1.6	2	5.2	5
Pb	µg/l	33	31 – 39	364	353 - 443

### References

1. Dtsch Arztebl 2002, 99: A3026-3029 [Heft 45]

2. Bundesgesundheitsbl-Gesundheitsforsch-Gesundheitsschutz 2005, 48, 246-250



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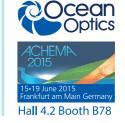
spectrometers. The spectral data captured with this setup is shown under Reason No. 3 above.

#### 4. Using the Flame is fun — and simple.

You spoke, we listened. Flame has new features to enhance the user experience, from interchangeable slits for measurement flexibility to device connectors for plug and play operation, simple coupling to external devices or integration into OEM devices. New LED indicators provide a quick visual reference for spectrometer status.



With the release of the "World's First Miniature Spectrometer," Ocean Optics helped to make spectroscopy portable, inexpensive and accessible. Now, inspired by your feedback, we've reinvented our miniature spectrometer to reflect the challenges of today's most demanding applications. Flame delivers greater thermal stability and lesser unit-to-unit variability, plus the freedom of interchangeable slits, simple device connectors and LED status indicators.



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